

# Material Science Callister Solution Manual

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42 For some hypothetical metal, the equilibrium number of vacancies at 900 C is  $23 \times 10^{25} \text{ m}^{-3}$  If the density and atomic weight of this metal are  $740 \text{ g/cm}^3$  and  $855 \text{ g/mol}$ , respectively, calculate the fraction of vacancies for this metal at 900 C Solution This problem is solved using two steps: (1) calculate the total number of lattice sites in silver,  $N$ , using

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**Chapter 2 Bonding - University of Tennessee**

Introduction To Materials Science, Chapter 2, Atomic Structure -Interatomic Bonding University of Tennessee, Dept of Materials Science and Engineering 8 Electrons in Atoms (II)  $\frac{3}{4}$ The quantum numbers arise from solution of Schrodinger's equation  $\frac{3}{4}$ Pauli Exclusion Principle: only one electron can have a given set of the four quantum numbers

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